

## PHY 342 HW Ch.4c

q4.6

For the first three lines in the Lyman series of the  $\text{He}^+$  ion, give the numerical values of the photon energies in eV, and the wave numbers in  $\text{cm}^{-1}$ . What is the limit of this series?

q4.7

(a) For what states ( $s, p, d, \dots$ ) does the wave function of the hydrogen atom vanish at the origin? (b) Briefly explain with classical physics arguments.

q4.8

Show that (a)  $[L_+, L_-] = 2\hbar L_Z$ ; (b)  $L^2 = L_- L_+ + \hbar L_Z + L_Z^2 = L_+ L_- - \hbar L_Z + L_Z^2$ ; (c)  $L_Z = -i\hbar \partial/\partial\varphi$  in spherical coordinates.

q4.9

The hydrogen atom is in state  $n = 3$ . What is the energy of the state in eV? What is the degeneracy? List all possible quantum numbers  $l$  and  $m$ .

q4.10

The wave function of a hydrogen atom is initially  $\Psi(t = 0) = A\psi_{200} - \frac{1}{\sqrt{2}}\psi_{300}$ . (a) Find  $A$  such that  $\Psi$  is normalized. (b) What are the probabilities of finding the atom in state  $n = 1, 2$ , and  $3$ ? (c) Graph (on a computer) the wave function  $\Psi(t = 0)$  using convenient units. Determine graphically where the maximum radial distribution is.

[extra credit] Animate the evolution of the wave function in time, using Python, for instance.